

# Baypure® - Chemical properties

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**Functional Chemicals  
BL Specialty Products**

Leverkusen, April 2008



**LANXESS**  
Energizing Chemistry

# Baypure® CX 100 (IDS) is a medium-strong chelant, Baypure® DS 100 (PASP) a weak one but to iron-III

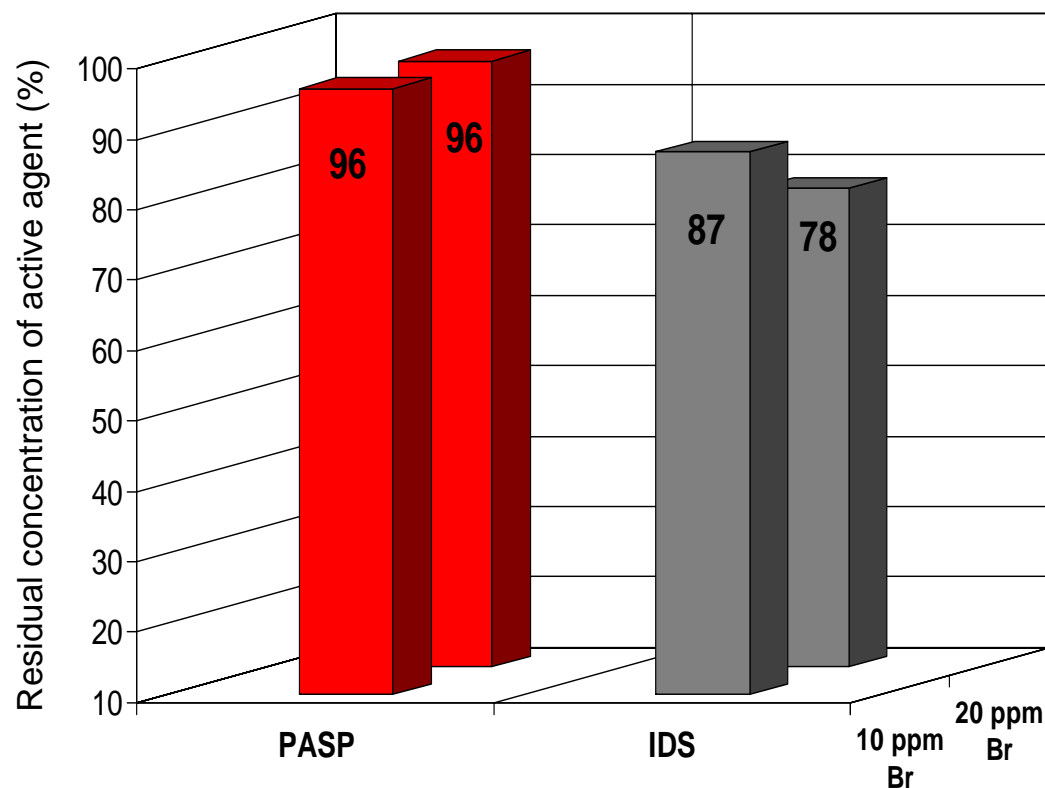
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Stability constants of complexes (log K values) at 20°C

Metal	Polyaspartic acid (PASP)	Iminodisuccinate (IDS)
Magnesium (Mg <sup>2+</sup> )	2.0	6.1
Calcium (Ca <sup>2+</sup> )	2,7	5.2
Strontium (Sr <sup>2+</sup> )		4.1
Barium (Ba <sup>2+</sup> )		3.4
Aluminum (Al <sup>3+</sup> )		14.1
Lead (Pb <sup>2+</sup> )		11.0
Chromium (Cr <sup>3+</sup> )	7.5	9.6
Manganese (Mn <sup>2+</sup> )	2.1	7.7
Iron (Fe <sup>2+</sup> )	10.0	8.2
Iron (Fe <sup>3+</sup> )	18.5	15.2
Cobalt (Co <sup>2+</sup> )		10.5
Nickel (Ni <sup>2+</sup> )		12.2
Copper (Cu <sup>2+</sup> )	4.8	13.1
Silver (Ag <sup>+</sup> )		3.9
Zinc (Zn <sup>2+</sup> )	2.2	10.8
Cadmium (Cd <sup>2+</sup> )	1.7	8.4
Mercury (Hg <sup>2+</sup> )	2.8	14.9

# Baypure® DS 100 and Baypure® CX 100 have a limited stability to bromine

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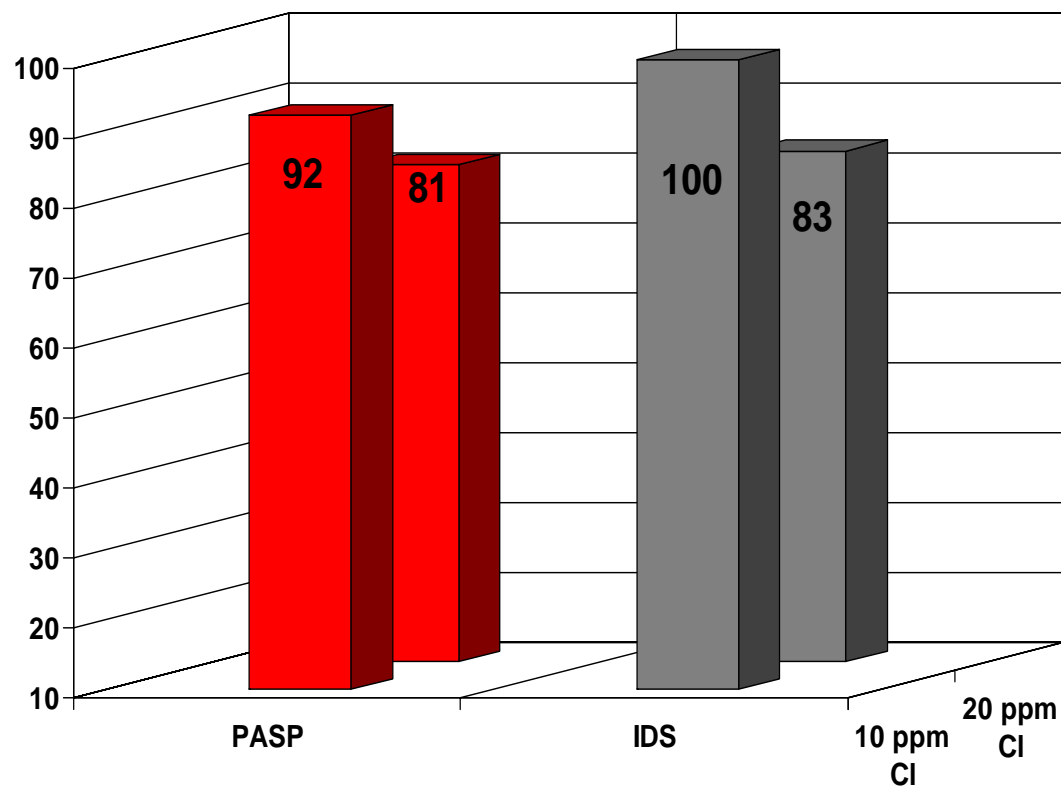


## Trial conditions:

Baypure DS 100/40% (PASP) and Baypure CX 100/34% (IDS), respectively, were dissolved to a concentration of 1 % of the commercial product in a water of a total hardness of 600 mg/l, calculated as calcium carbonate. The dosages of the halogens, chlorine and bromine, were at 10 and 20 mg/l, respectively. The samples were stored for two hours at a temperature of 40°C

# Baypure® DS 100 and Baypure® CX 100 have a limited stability to chlorine

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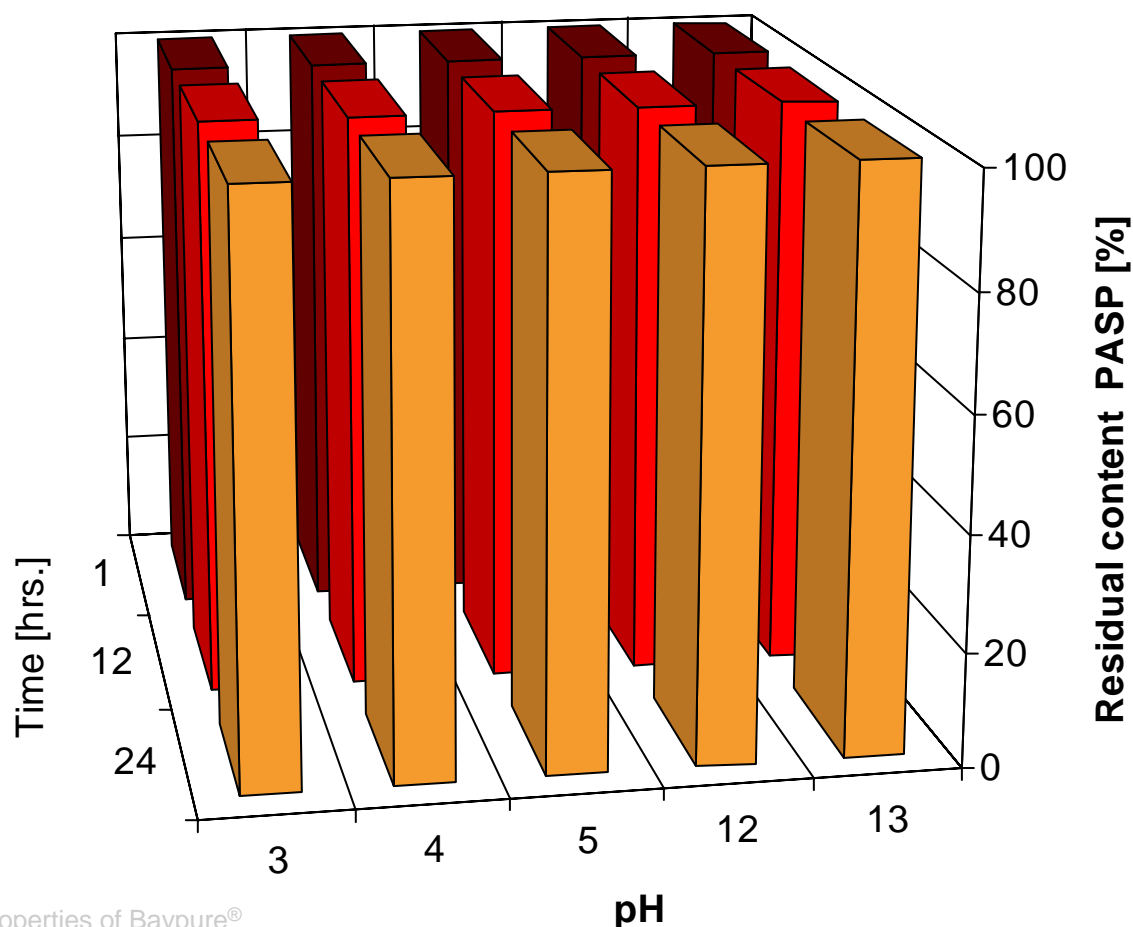


## **Trial conditions:**

Baypure DS 100/40% (PASP) and Baypure CX 100/34% (IDS), respectively, were dissolved to a concentration of 1 % of the commercial product in a water of a total hardness of 600 mg/l, calculated as calcium carbonate. The dosages of the halogens, chlorine and bromine, were at 10 and 20 mg/l, respectively. The samples were stored for two hours at a temperature of 40°C

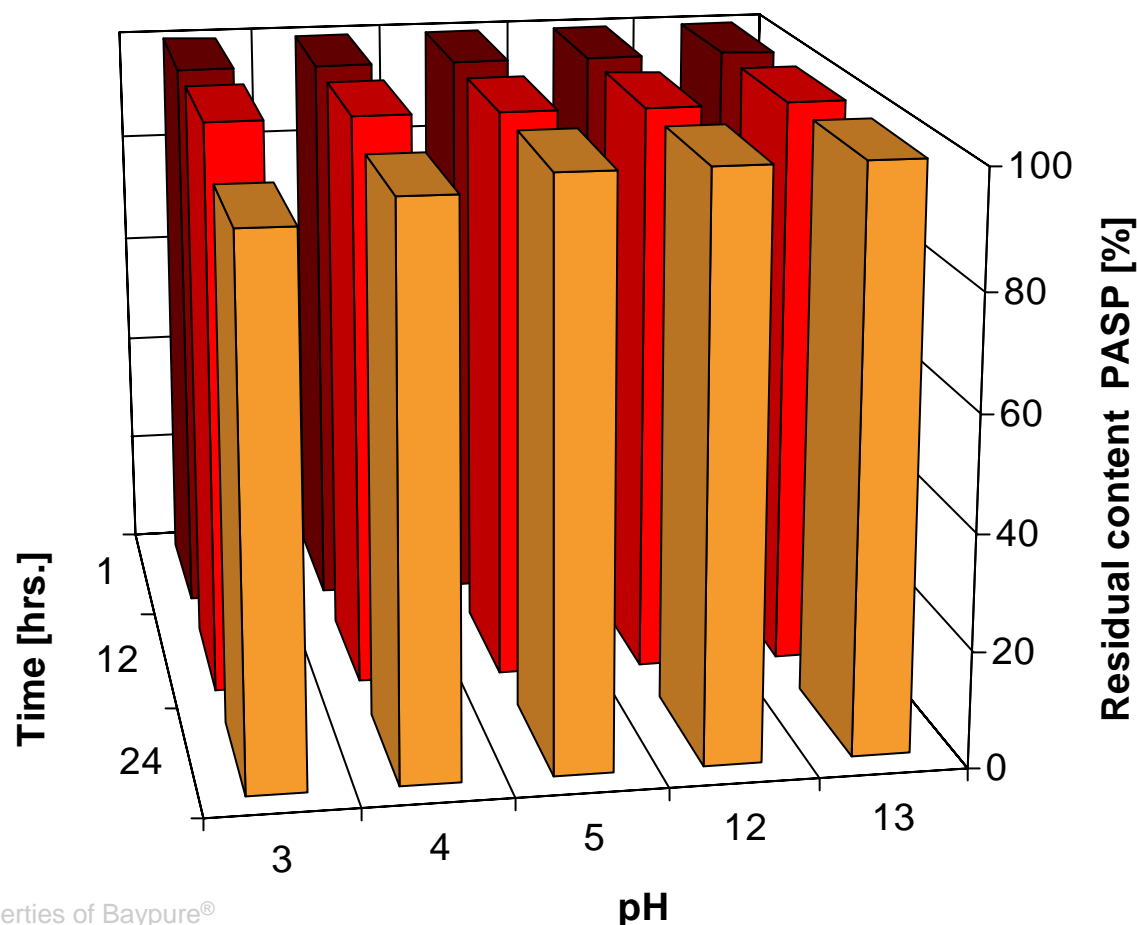
# Baypure® DS 100 has a very good stability under typical cleaning formulation conditions

Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 20 °C in a 3 % aqueous solution at different pH values



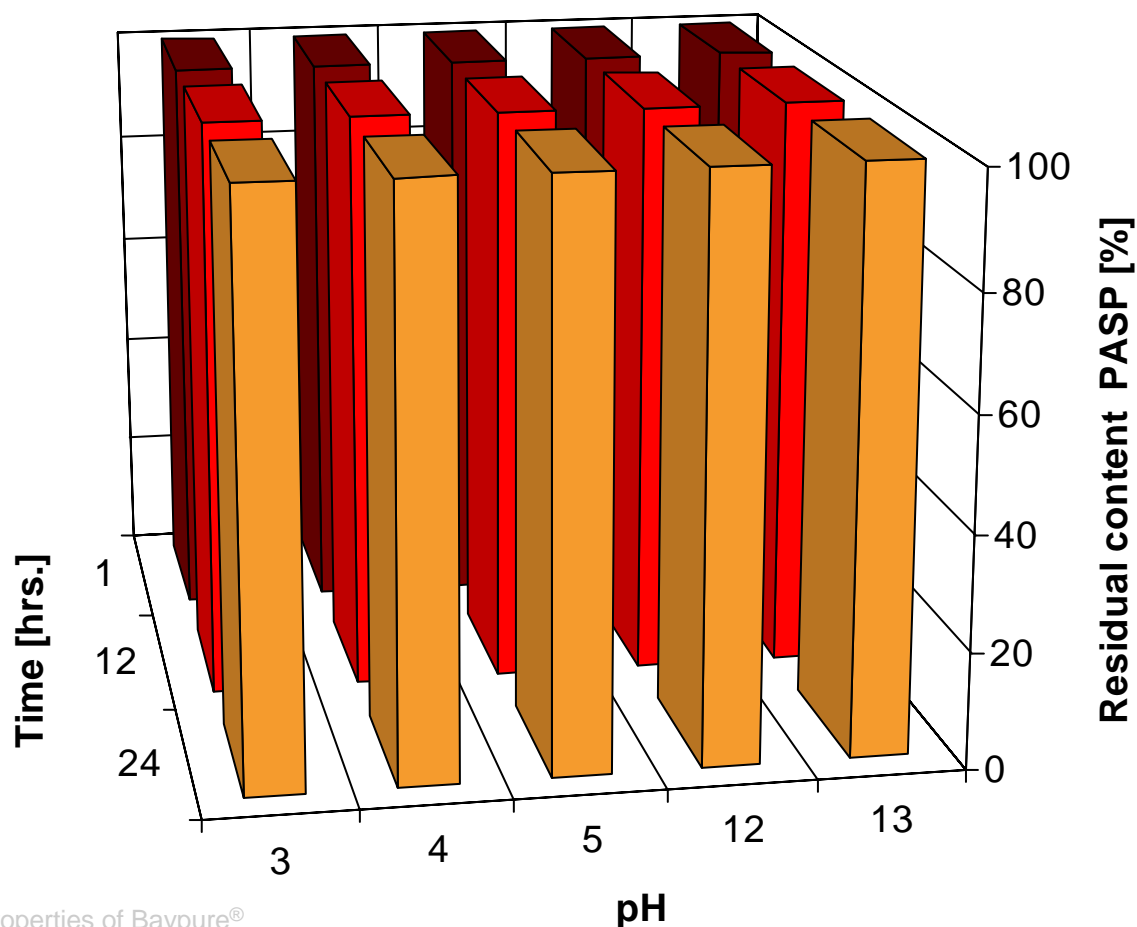
# Baypure® DS 100 has a very good stability under typical cleaning formulation conditions

Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 50 °C in a 3 % aqueous solution at different pH values



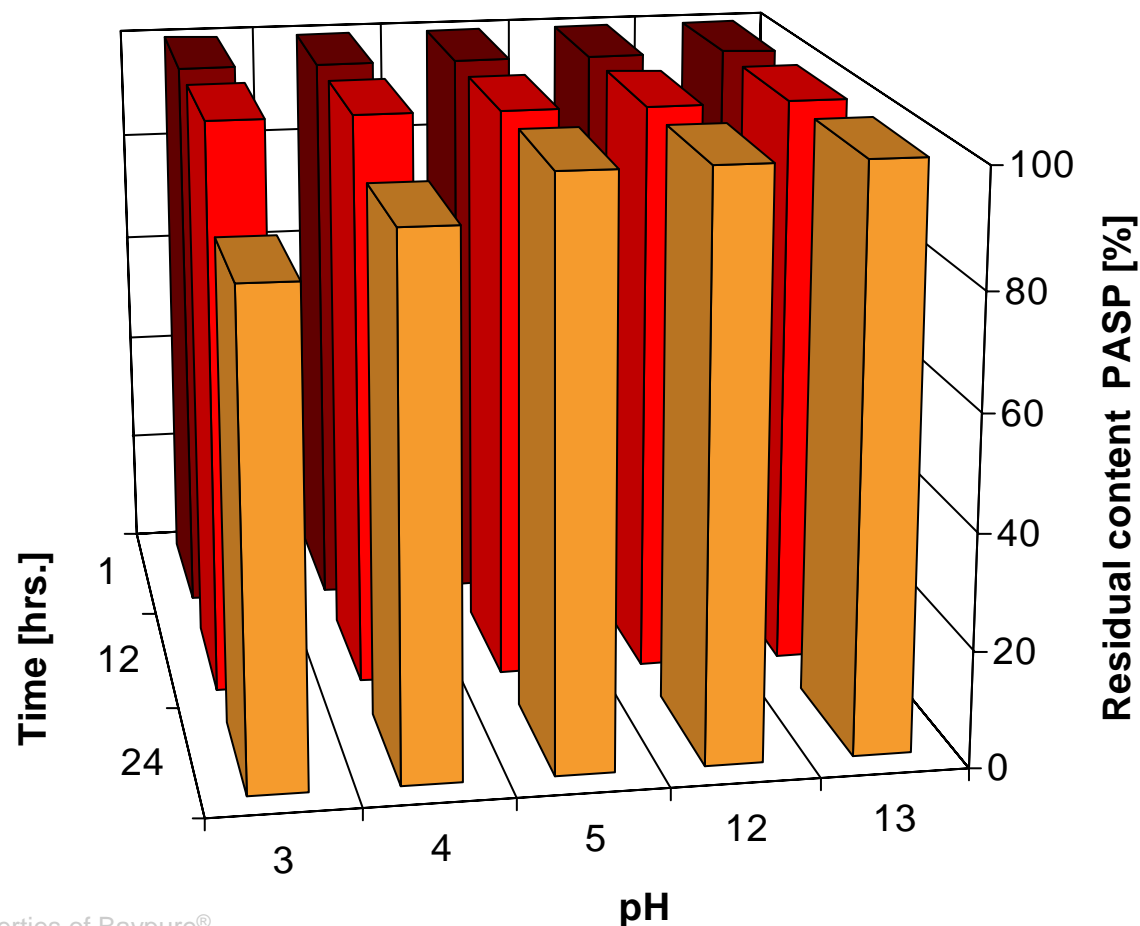
# Baypure® DS 100 has a very good stability typical cleaning formulation conditions

Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 20 °C in a 5 % aqueous solution at different pH values



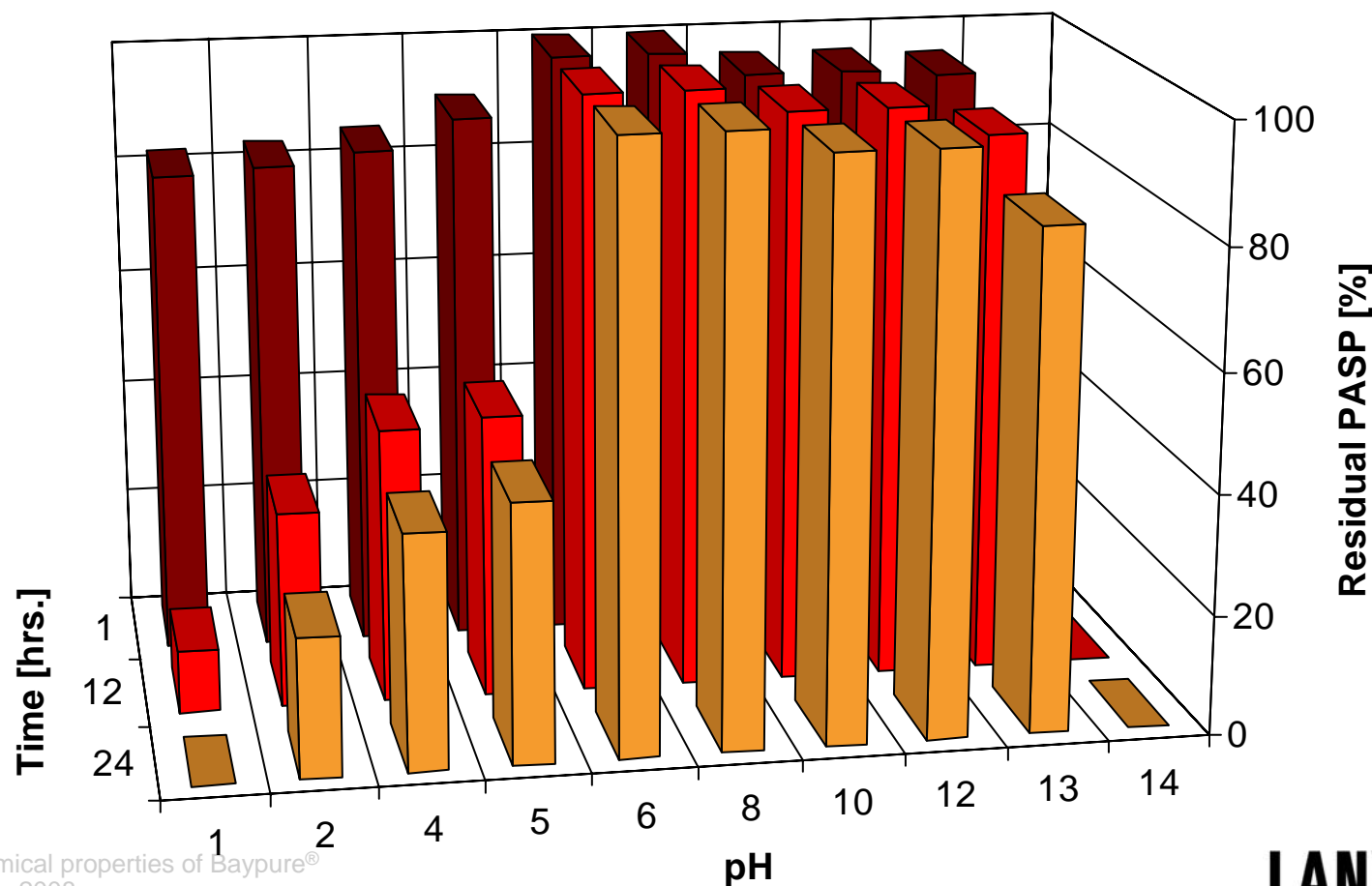
# Baypure® DS 100 has a very good stability under typical cleaning formulation conditions

Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 50 °C in a 5 % aqueous solution at different pH values



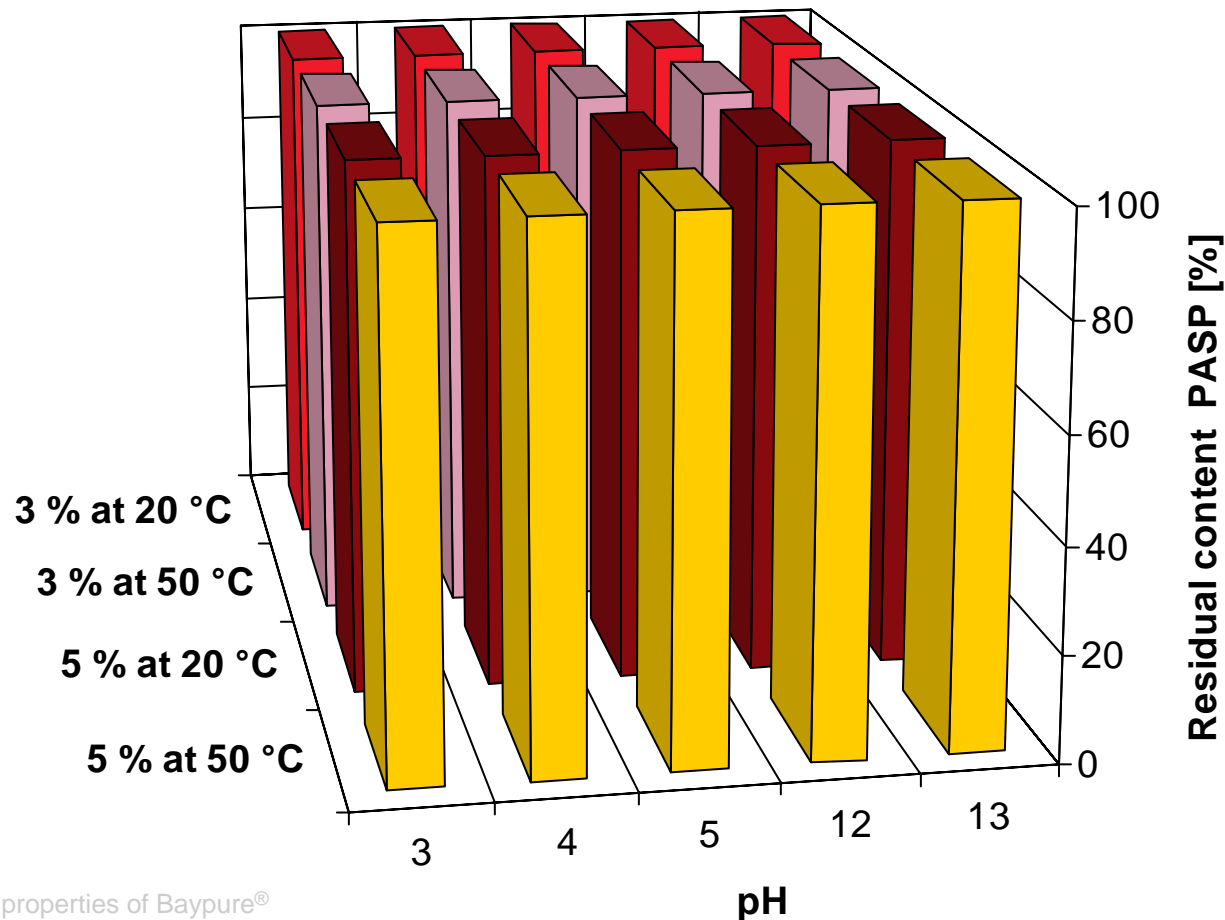
# Baypure® DS 100 has a sufficient stability under extreme cleaning formulation conditions

Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 95 °C in a 1 % aqueous solution at different pH values



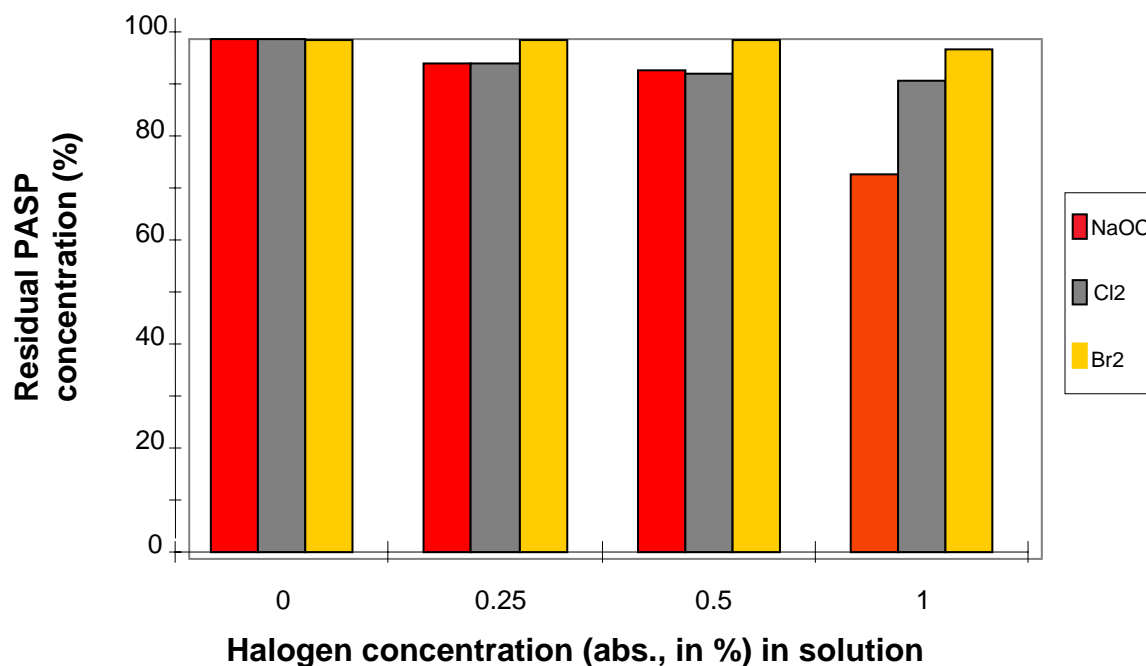
# Baypure® DS 100 has a very good stability under typical cleaning formulation conditions

**Summary:** Thermal stability of Polyaspartic acid sodium salt (active ingredient in Baypure® DS 100) at 20 and 50 °C, respectively, in a 3 or 5 % aqueous solution at different pH values



# Baypure® DS 100 is reasonably stable to bromine, but shows a limited resistance to chlorine

Stability of Polyaspartic acid sodium salt (PASP) to bromine, chlorine, or sodium hypochlorite at a temperature of 20°C and after 24 hours

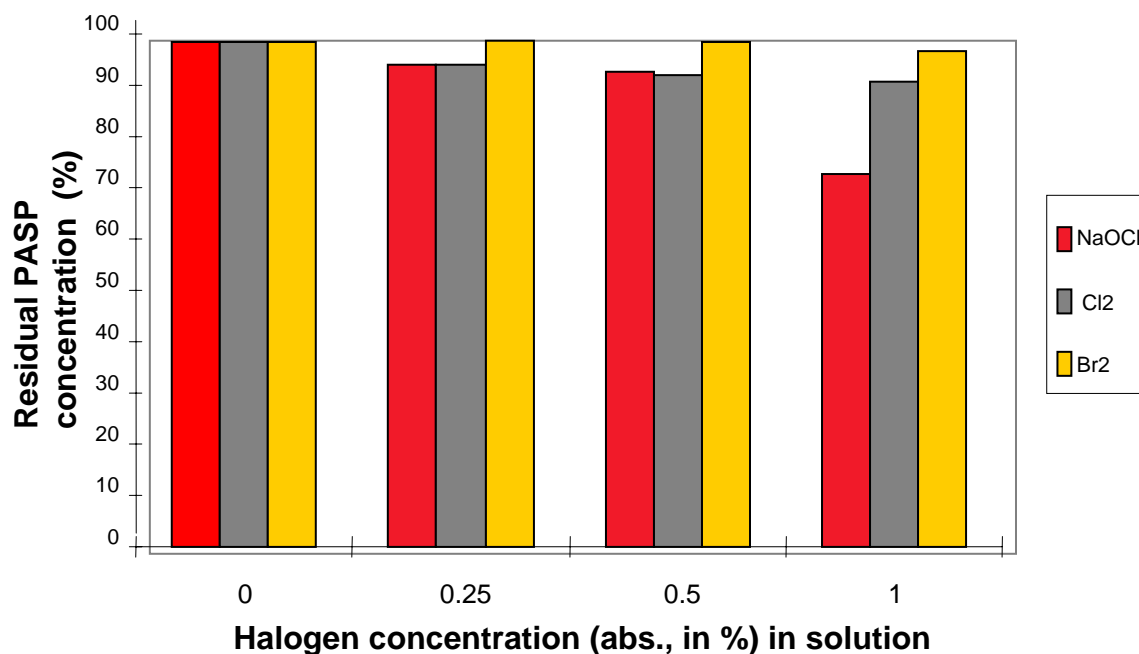


## Trial conditions:

Baypure® DS 100/40% (PASP) was dissolved in water (1 % solution of active agent) and mixed with a 1 % solution of chlorine or bromine, respectively, in synthetical tap water (total hardness: 250 mg/l as calcium were dissolved to a concentration of 1 % of the commercial product in a water of a total hardness of 600 mg/l, calculated as calcium carbonate. The dosages of the halogens, chlorine and bromine, were at 10 and 20 mg/l, respectively. The samples were stored for two hours at a temperature of 20°C

# Baypure® DS 100 is reasonably stable to bromine, but shows a limited resistance to chlorine

Stability of Polyaspartic acid sodium salt (PASP) to bromine, chlorine , or sodium hypochlorite at a temperature of 50°C and after 24 hours

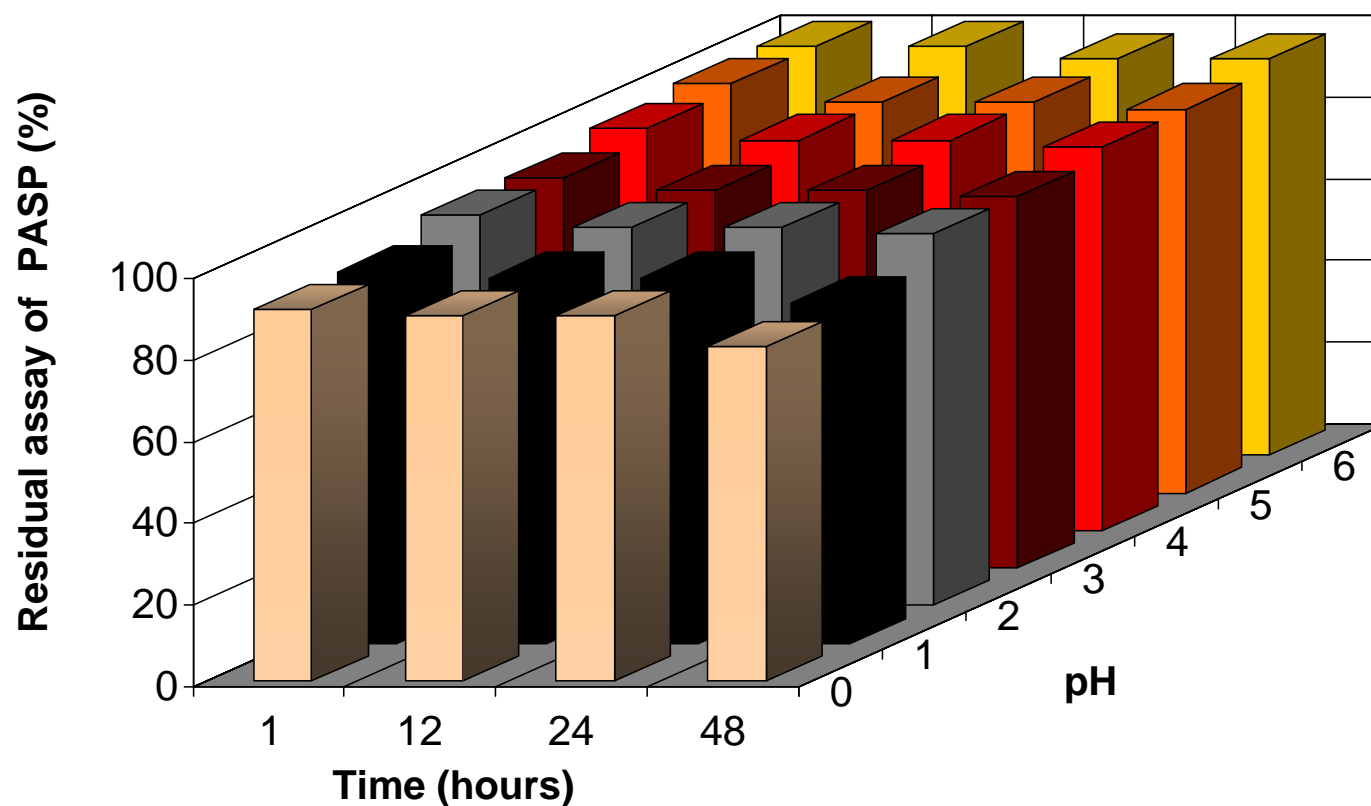


## Trial conditions:

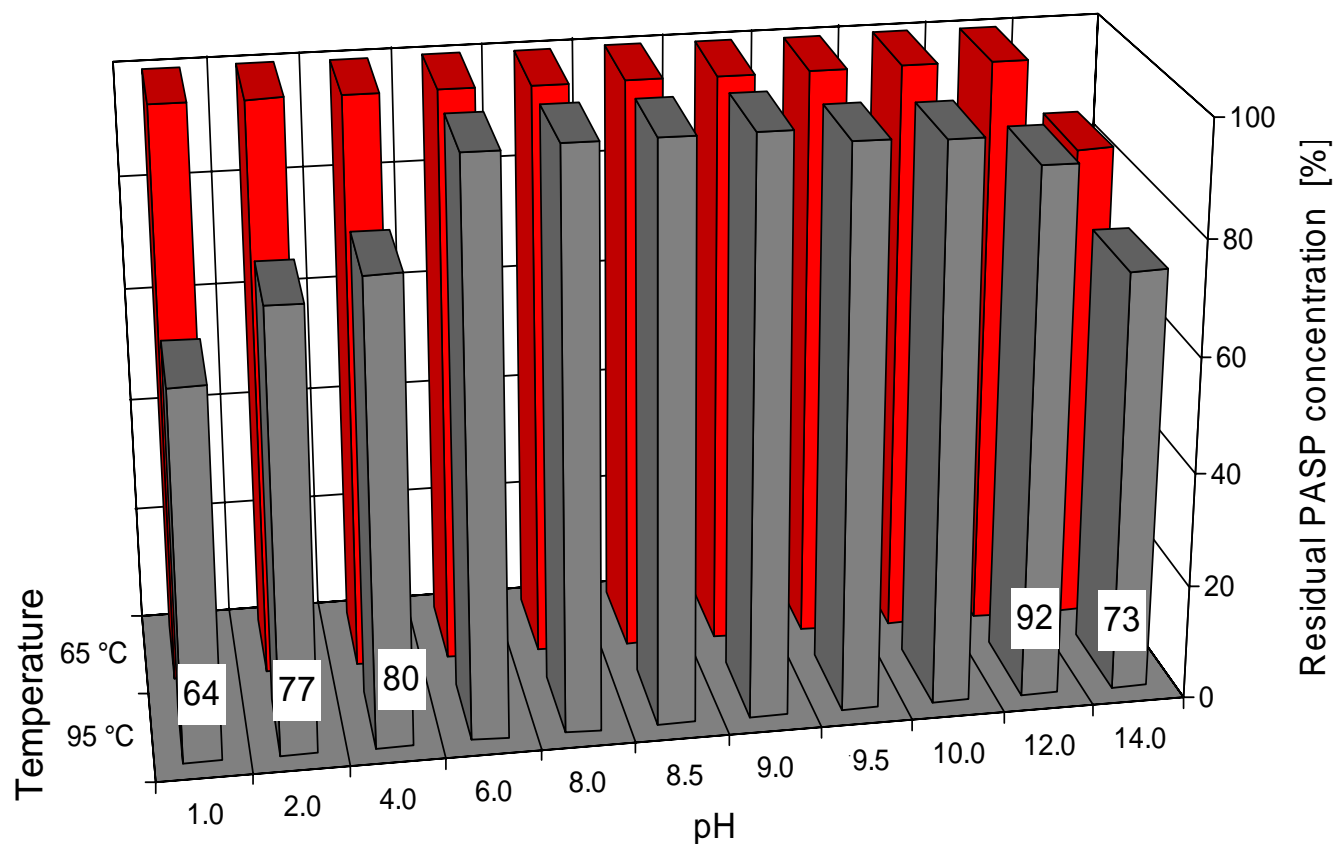
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# Baypure® DS 100 (Polyaspartic acid sodium salt, PASP) is stable to chlorine dioxide in acidic solution

Stability of PASP to chlorine dioxide at pH 0 to 6 and a temperature of 40 °C



# Baypure® DS 100 is stable to hydrogen peroxide in most conditions of application



## Trial conditions:

Baypure® DS 100/40% (PASP) was dissolved in water (1 % solution of active agent) and mixed with an aqueous solution of hydrogen peroxide (in total 4 mol equivalents of  $\text{H}_2\text{O}_2$  versus PASP). The mixture was kept at 65 or 95°C, respectively, for one hour. After the cool-down, the residual assay of PASP was determined by complexometric titration with  $\text{FeCl}_3$  solution.